



PULPING OF DIFFERENT RAW MATERIALS (WOOD AND AGRO WASTE)

Dr. Shuank Malik Cellulose and Pap er Discipline ICFRE- Forest Rese arch Institute Dehradun



Brief History of Paper Making

- **Papyrus in Ancient Egypt.**
- True papermaking began in China (~100 AD).
- Spread to Middle East and Europe (by the 15th century).
- Arrival in North America (1690), near Philadelphia.
- First Indian mechanized paper mill: 1812, Serampore, West Bengal.









Key Milestones in Pulping Process

1840	 Ground wood pulping method de Germany – came into commercia 1860s.
1854	• First Manufacture of pulp from when the soda process (England).
1867	 Patent issued to Benjamin Thigh for the sulphite pulping process- commercially in 1874.
1884	• 1884 - Kraft pulping process in Carl Dahl (Germany).

eveloped in use in the

wood using

man (USA) - first used

nvented by

INTRODUCTION



FIBRE CHEMISTRY

- Plant fibres are composed mainly of cellulose, hemicelluloses, lignin and extractives.
- These components influence pulp yield, strength, brightness, and reactivity.
- Their ratios vary between woods and agrowastes, softwoods and hardwoods.





Cellulose

- Linear homopolysaccharide of β-1,4linked D-glucose units
- Repeating unit: Cellobiose (2 glucose molecules)
- Degree of polymerization (DP):Wood pulp: 600–1500, Cotton: up to 10,000+
- Insoluble in water and most solvents
- Forms hydrogen bonds gives strength, flexibility
- Exists in crystalline and amorphous regions.



Hemicellulose

- Branched heteropolysaccharide mix of sugars: Hexoses: glucose, mannose, galactose and Pentoses: xylose, arabinose
- Less crystalline → more hydrolyzable

Functions:

- Interacts with cellulose and lignin
- Affects water retention, pulp flexibility, bonding strength
- More abundant in hardwoods (xylan) than softwoods (mannan)



Lignin

- Complex 3D polymer of phenylpropanoid units: Syringyl (S), Guaiacyl (G), Hydroxyphenyl (H).
- Provides rigidity, water resistance.
- Cross-links with cellulose–hemicellulose matrix.
- Must be chemically removed during pulping (especially kraft/sulphite).
- Source of black liquor in kraft process.



Extractives

- Include resins, fats, waxes, terpenes
- **Extracted using water/organic** solvents

Affect:

- **Pulp Brightness**
- Chemical Consumption
- **Paper Odor and Aging**
- **More abundant in agro-wastes** and softwoods.





PULP AND PAPER MANUFACTURING PROCESS



RAW MATERIAL PREPARATION

- Essential for efficient chemical penetration and uniform pulping
- Involves processing wood or agroresidues into suitable forms
- Main steps:
- 1.Bark removal
- 2. Chipping (or cutting)
- **3.Screening and cleaning**



PULPING AND ITS TYPES

- Pulping is the process of converting fibrous raw materials into a fibrous mass.
- It can be achieved through mechanical, thermal, chemical, or a combination of these methods.

Chemi – mechanical Pulping Mechanical Pulping

PULPING TYPES

Chemical Pulping

Semichemical Pulping

MECHANICAL PULPING

- Uses physical force to separate fibres.
- High yield, low strength.
- Suitable for newsprint, paperboard, etc.





TYPES OF MECHANICAL PULPING

Stone Groundwood Pulping (SGW)

Refiner Mechanical Pulping (RMP)

Thermo-mechanical Pulping (TMP)

Chemi-thermo-mechanical Pulping (CTMP)

Advantages and Disadvantages of Mechanical Pulping

Advantages

- High fibre yield (up to 95%)
- Lower raw material cost.
- Good bulk, opacity, and printability.
- Suitable for low-cost paper products.

- •Weak fibres \rightarrow poor paper
- strength.
- Paper discolours (yellows) in light.

Disadvantages

• High energy consumption.

CHEMICAL PULPING

- Chemicals dissolve lignin.
- Lower yield but stronger pulp.
- Ideal for writing, printing, packaging paper.



Kraft pulping

Soda pulping

Sulphite pulping

TYPES OF CHEMICAL PULPING

SODA PULPING

- Uses only NaOH
- Ideal for agro-residues: bagasse, straws and grasses
- No sulphur emissions
- Limitations:
- **1.Lower strength than kraft**

2.No recovery cycle for agrowastes





KRAFT PULPING

- Also called Sulphate process.
- Uses NaOH + Na₂S (white liquor).
- Most widely used method globally.
- Pulp: strong, brown, versatile.
- <u>Recovery system:</u>
- **1.Spent black liquor → recovery boiler**
- $2.Na_2SO_4 \rightarrow reduced to Na_2S$
- <u>Applications</u>:

Packaging, printing paper, tissues.



SULPHITE PULPING

- Uses H₂SO₃ and bisulphite salts Advantages (Ca, Mg, Na, NH_4)
- Operating pH defines type:
- 1. Acid Sulphite (pH 1–2)
- 2. Bisulphite (pH 3–5)
- Produces light-coloured, easy-tobleach pulp
- Less suitable for bark, resinous wood
- Applications: printing paper, tissue, dissolving pulp

of Process

- Produces strength pulp
- Utilizes
 - technology for chemical recove
- Handles wide value wood species
- Tolerate bark pulping process

Kraft	Advantages	of sulphite	
	process		
highest	• Produces	brighter	
	unbleached pulp		
proven	• Pulp is easier to bleach		
efficient	to full brightness		
ry	 Produces 	higher yield	
ariety of	of bleached pulp		
	• Pulp is easi	ier to refine	
in the			

PROCESS CHEMICALS WOOD TYPE YIELD

Kraft	NaOH + Na₂S	All	40-50%
Sulphite	H ₂ SO ₃ + HSO ₃ ⁻	Selected species	45-55%
Soda	NaOH	Agro- waste	45-55%

SUMMARY OF CHEMICAL PULPING

SEMI-CHEMICAL PULPING

- Combines chemical softening + mechanical refining
- Partial lignin removal
- Used mainly for corrugated medium (fluting)
- Key methods:
- 1. NSSC (Neutral Sulphite Semichemical): Hardwood, buffered Na₂SO₃ + Na₂CO₃ Yield: 65–80%
- 2. High-Yield Kraft/Sulphite:

Limited cooking, mechanical defibering **Yield: 55–75%**



	Kraft	Acid Sulphite	Bisulphite	NSSC
Chemicals	NaOH Na ₂ S	H_2SO_3 $M(HSO_3)$ $(M = Ca,$ $Mg, Na,$ $NH_4)$	M(HSO ₃) (M=Mg, Na, NH ₄)	Na ₂ SO ₃ Na ₂ CO ₃
Cooking Time	2-4 h	4-7 h	2-4 h	1/4 – 1 h
Liquor pH	13+	1-2	3-5	7-9
Cooking Temp	170-180 °C	120-135 °C	140-160 °C	160-180 °C

COMPARISON OF **CHEMICAL &** SEMI-CHEMICAL PULPING

Pulping Comparison Processes **Najor**

S.No.	Classification	Process Name
1.	Mechanical	Stone Groundwood
		RMP
		TMP
2.	Chemi-mechanical	Chemigroundwood
		Cold Soda
3.	Semichemical	NSSC
		High Yield Sulfite
		High Yield Kraft
4.	Chemical	Kraft
		Sulfite
		Soda

Wood Used	Form of Wood	Yield(%)
Softwood (mostly)	Bolts	90-95
Softwood (mostly)	Chips	90-95
Softwood	Chips	90
Hardwood	Bolts	85-90
Hardwood	Chips	85-90
Hardwood	Chips	65-80
Softwood (mostly)	Chips	55–75
Softwood (mostly)	Chips	50-70
Both	Chips	40-50
Both	Chips	45-55
Hardwood	Chips	45-55

DISSOLVING PULP

- High-purity chemical pulp **Used for:**
- 1. Rayon
- 2. Cellulose acetate
- 3. Cellophane
- 4. CMC (carboxymethyl cellulose)

Made from:

- **1. Modified kraft/sulphite pulping**
- **2.** Cotton linters
- High alpha-cellulose (>90–95%)
- Viscose process: converts pulp into fibres or films



COMMON TERMS ASSOCIATED TO PULPING

- <u>DELIGNIFICATION</u>- It is the process of breaking down the chemical structure of lignin and rendering it soluble in a liquid.
- **<u>KAPPA NUMBER</u>** It is a measure of the lignin content of the pulp.
- **<u>PULP VISCOSITY</u>** It is a measure of the average chain length of cellulose.
- FIBRE LIBERATION POINT It occurs when sufficient lignin has been removed during pulping that wood chips will be soft enough to break apart into fibres with little or no mechanical action.
- **FULL CHEMICAL PULP-** A pulp produced by chemical methods only.
- DISSOLVING PULP-It is a low yield (30-35%) bleached chemical pulp that has a high cellulose content.
- <u>LIQUOR</u>- Cooking or pulping liquor is an aqueous solution of chemicals used for delignification of wood for pulping.
- <u>H-FACTOR</u> It is a pulping variable that combines cooking temperature and time into a single variable that indicates extent of delignification.

COMMON TERMS ASSOCIATED TO PULPING • WHITE LIQUOR- It is a fresh pulping liquor for the kraft process consisting of the active

- pulping species, NaOH and Na2S.
- **BLACK LIQUOR-** It is the waste liquor from the kraft pulping process after pulping is completed. It contains most of the original cooking inorganic elements and the degraded dissolved wood substance.
- GREEN LIQUOR- It is partially recovered form of kraft liquor. It is obtained after the combustion of black liquor in the recovery boiler.
- **TOTAL CHEMICAL OR TOTAL ALKALI**. It is the sum of all sodium salts in the liquor (as Na_2O).
- **ACTIVE ALKALI** It includes the active ingredients in the pulping process i.e. NaOH and Na₂S.
- **<u>EFFECTIVE ALKALI-</u>** It is the ingredients that will actually produce alkali under pulping conditions.
- **SULPHIDITY** It is the ratio of Na₂S to active alkali.
- **CAUSTICITY-** It is the ratio of NaOH to active alkali expressed as %.
- **RESIDUAL ALKALI** The EA remaining in a cook at its completion.
- **DEAD LOAD-** Inorganic materials, principally NaCl, present in kraft liquor, but cannot be regenerated in the recovery process.



FACTORS **AFFECTING YIELD** DURING PULPING

B N N S N S N S S S S S S



It is done to remove air filled inside the pores in the chips. This eases the process of chemical penetration.

> It is the process by which cooking liquor impregnates the chips and is equally distributed within the chips.

> > Bulk penetration is the diffusion of molecules or ions of the cooking liquor while diffusion is the movement of molecules from a higher concentration to a lower concentration and doesn't occur until the pores are filled with liquor.

It occurs at a temperature of about 165-170°C.

It refers to the release of pressure after cooking completes and the delignified pulp is taken out of the digester.

OTHER PULPING METHODS







CONCLUSION



- Papermaking begins with the conversion of plant-based raw materials into pulp.
- The pulp is prepared using mechanical, chemical, or semichemical processes, depending on the desired paper quality.
- Key components of fibre—cellulose, hemicellulose, and lignin influence pulp behaviour and paper properties.
- After pulping, the pulp is cleaned, refined, formed into sheets, pressed, and dried to make paper. The selection of raw materials and pulping method affects paper strength, brightness, durability, and cost.
- Sustainable practices, including use of agro-waste and chemical recovery systems, are vital for eco-friendly papermaking.



<u>CONTACT US</u>

0135-222-4386

<u>smalik@icfre.org;</u> <u>shuankamlik01@gmail.com</u>

CELLULOSE AND PAPER DISCIPLINE FOREST PRODUCTS DIVISION ICFRE- FOREST RESEARCH INSTITUTE DEHRADUN (248006), UTTARAKHAND

