Standardisation of Pulp Mill Effluent for Irrigation Purposes

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SUMMARY

The purpose of this study was to standardise water of irrigation quality by mixing grade III effluent from pulp mill and chlorination and hypochlorite stage bleach plant effluent.

Total dissolved inorganic solids, chlorides, sulphates, boron and percent sodium are important parameters which have to be taken into consideration in deciding the usefulness or otherwise of waste water for agricultural purposes.

It was observed that when 2 volume of grade III effluent and one volume of chlorination and hypochlorite effluent from bleach plant were mixed, all the above mentioned parameters come within the limits prescribed in IS : 3307–1965 (Tolerance limits for discharge of industrial effluent on land for irrigation).

INTRODUCTION

Industries are the backbone of economical development of any country. But one of the common drawback of industrialisation is the production of wastes from all most all industries which pollute the environment. Waste treatment and disposal is a big challange before the industries. The best solution of this problem is the utilization of waste materials from the industries.

Pulp and paper industry is one of the largest industry in India today, which consumes huge quantity of water (40,000 to 100,000 gallons of water per ton of paper). As the product does not carry any water, most of the water used in the process reappears as wastewater. Usually this water is discharged into the stream, which pollute the natural water resources. Hence to save our water resources and to permit the industry to expand the discharge of wastewaters into streams should be controlled. If this wastewater can be used for crop irrigation this will serve the nation in two ways. First it will solve the wastewater disposal problem of the industry and second it will give much needed irrigation water for agricultural purposes.

Sewage effluents and domestic wastewaters have

*Senior Chemist, **Chemist, Research Division, Orient Paper Mills, Amlai (M. P.) been in use in many places and have proved quite satisfactory for crop irrigation. But regarding the utilization of industrial effluents for land irrigation not much information is available. Since the industrial wastewaters may present more of a hazard to plant life due to their higher concentration of soluble salts, much attention has not been paid towards their utilization for crop irrigation. But looking to the limited water resources and increasing industrial pollution, serious attention will have to be paid towards re-utilization of industrial wastewater either for land irrigation or for the industry itself.

Efforts have been done by some paper industries in India to utilize their wastewater for irrigation purpose. Seshasayee Paper and Boards Ltd., Madras is using part of the combined effluent for paddy irrigation, Pudumjee Pulp and Paper Mills Ltd., Maharashtra is supplying their wastewater (after treatment) to farmers for crop irrigation and some other paper mills tried their effluent for Eucalyptus irrigation.

Present investigation have aimed to standardise a mixture of different type of wastewaters coming out from the pulp mill for irrigation purpose. In this study efforts have been made to standardise irrigation water as per IS : 3307-1965 (Tolerance limits for discharge of industrial effluents on land for irrigation).

MATERIAL AND METHOD

Orient Paper Mills, Amlai is an integrated pulp

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and paper mill producing 200 tons of paper per day, using sulphate process. The mill requires about 14 million gallons of fresh water and discharges about 11.5 million gallons of wastewater per day. Wastewater discharged from different sections of the mill have been segregated into three grades:

- Grade I— is cooling water which is practically uncontaminated and is reused in the process.
- Grade II— this waste is a mixture of wastewater from paper machine, chlorination and hypochlorite stage bleach plant effluent and wastewater from Chipper house. Total volume of this waste is about 7.0 million gallons per day. Out of which 3 to 3.5 million gallons of wastewater is from chlorination and hypochlorite stage from bleach plant.
- Grade III is a mixture of wastewater from pulp washing, digester and recovery sections floor washes, and caustic extraction effluent from bleach plant. The volume of this effluent is 3.5 to 4.0 million gallons per day. This effluent is dark brown in colour with high pH and rich in sodium.

Wastewater from paper machine section is less contaminated and it can be reused in the process itself after some treatment. Hence the present study is concentrated on the acidic effluent of bleach plant (chlorination and hypochlorite stage) and Grade III effluent.

Composite samples of untreated Grade III effluent and chlorination and hypochlorite stage bleach plant effluent were collected on various dates and analysed in the laboratory as per "Standard method for water and wastewater analysis" 12th edition published by APHA and "Text book of Quantitative Analysis" by A.I. Vogal. The parameters studied were—pH, EC (electrical conductivity), BOD, COD, dissolved solids, chloride, sulphate, carbonate, bicarbonate, calcium, magnesium, sodium and potasium. Results are presented in Table-I.

After characterising both the wastewater, Grade III effluent was mixed with chlorination and hypochlorite stage bleach plant effluent in the ratio of 3:1, 2:1 and 1:1, and analysed for the above mentioned parameters. Analysis results are given in Table-III A, B, C and D.

RESULTS AND DISCUSSIONS

Quality of the irrigation water is evaluated on the basis of quantity of dissolved salts and its composition. Salinity or alkali hazards that are likely to develop,

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depends upon the quality of irrigation water used. The generally accepted criteria for irrigation water and their potential hazards to crop growth are:

1. SALINITY

It comprises due to the high concentration of total soluble salts mostly inorganic in nature. High salinity retards Plant growth, producing smaller plants with fewer and smaller leaves. It is generally expressed in terms of electrical conductivity (E.C.) water with the E.C. of 750-2250 micromohs/cm are widely used for irrigation to get satisfactory crop growth. But water having E.C. above 2250 micromohs/cm are seldom used for crop irrigation.

2. SODICITY (PROPORTION OF SODIUM TO THE OTHER CATIONS)

High sodium percent in irrigation water is harmful to the soil as well as to the Plants. High sodium percent reduces permeability of soil and deteriorate soil structure. Sodium rich water also effect the Plants, such as leaf burning and defaliation have been observed. The sodium percent in irrigation water should be below 60.

3. BORON

For the normal growth of the plant Boron is found to be an essential plant nutrient. But it is required in traces. Boron is found to be very toxic to certain plant species. Boron should be less than 2 mg/l in irrigation water.

4. CARBONATE AND BI-CARBONATE

Presence of appreciable quantity of carbonate and bi-carbonate in irrigation water causes the precipitation of Calcium (Ca) and Magnesium (Mg) as their carbonates and thereby increases the Sodium (Na) percent of the soil water.

Looking to the analysis results presented in Table-I it becomes clear that chlorination and hypochlorite stage bleach plant effluent alone can not be used for irrigation purpose, since it is acidic in nature with pH less than 3.0 and chlorides more than 1000 mg/l. Whereas permissible limits as per I.S: 3307-1965 is pH 5.5 to 9.0 and chlorides less than 600 mg/l. Work on the utilization of Grade III effluent for agriculture purposes has been done by NEERI-Nagpur at Amlai. Although their findings has not been published. But the characteristics of the effluent shows that it can not safely be used for crop irrigation, as it is highly alkaline having pH upto 10.0 and percent sodium is as high as 84, where as permissible limit is only 60 percent.

Results presented in Table-III A, B, C and D

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TABLE—1. CHARACTERISTICS OF GRADE III EFFLUENT AND CHLORINATION AND HYPOCHLORITE EFFLUENT FROM BLEACH PLANT

Parameters	Grade III effluent	Chlorina- tion and hypochlorite effluent
pH E.C. milimohs/c.m. BOD ₅ at 20 °C mg/l COD ,, Dissolved solids ,, Chlorides (as Cl) ,, Sulphates (as SO ₄) ,, Alkalinity (as CaCO ₃) ,, Calcium (Ca) ,, Magnesium (Mg) ,, Sodium (Na) ,, Potassium (K) ,, Percent Sodium	8.6-10.0 1.5-2.2 180-360 1000-2000 1200-2800 100-300 45-150 200-600 50-200 12-60 300-600 20-40 55-85	$\begin{array}{r} 1.6-3.5\\ 3.0-5.8\\ 80-160\\ 500-800\\ 1700-3000\\ 800-1500\\ 35-60\\ 200-450*\\ 600-1100\\ 100-250\\ 110-180\\ 10-20\\ 16-21.5 \end{array}$
S.A.R.	55–85 8–18	1.5-2.0

* Refers for acidity.

TABLE—II. TOLERANCE LIMITS FOR DIS-CHARGE OF INDUSTRIAL EFFLUENTS ON LAND FOR IRRIGATION (IS : 3307-1965)

Characteristics			Tolerance limits
1. pH		-i	5.5-9,0
2. Dissolved solids	mg/l	Max.	2100
(inorganic)			
3. Sulphates (as SO_4)	,,	,,	1000
4. Chlorides (as Cl)	,,	,,	600
5. Percent sodium		,, ·	60
6. BOD ₅ at 20 $^{\circ}$ C	mg/l	,, .	500
7. Oil & Grease	,,	,,	30
8. Boron (as B)	,,	,,,	. 2

reveals that when Grade III effluent and bleach plant effluent are mixed in the ratio of 2:1 all the required parameters comes within the limits prescribed for irrigation water. pH of the water lie in between 5.6 to 8.0, E.C. 2220 to 2290 micromohs/cm., dissolved solids 1800 to 2100 mg/l, chlorides 532.5 to 610 mg/l, sulphates 60 to 70 mg/l and sodium 43.8 to 55.6 percent. Comparing these results with the specifications presented in Table-II (As per IS : 3307-1965) it becomes clear that all the parameters are within the tolerance limits for discharge of industrial effluents on land for irrigation.

Boron has not been determined since neither the raw-material used nor the chemicals employed in the process contain Boron.

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 эН	Electrical con- ductivity E.C.	BOD ₅	COD	Dissolved solids	Chlorides (as Cl)	Sulphates (as SO ₄)	Carbonate (as CaCo3)	Bi-Carbonate (as CaCo3)	Calcium (as Ca)	Magnesium (as Mg)	Sodium (as Na)	Potasium (as K)	Percent Sodium	S.A.R.
	Milimohs/cm	mg/l	mg/l	mg/l	mg/l	mg/1	mg/l	mg/l	mg/l	mg/1	mg/l	mg/l		16
2	3	4	5	6	7	8	9	10	11	12	13	14	15	10
9.4 7.6 7.0 6.0 2.8	1.59 2.13 2.22 2.49 4.20	220 190 190 180 130	1232 1072 1000 1000 572	1600 « 1750 1800 1800 2000	276.9 454.4 532.5 646.1 1065	65 75 60 50 35	5 Nil Nil Nil Nil	250 180 10 Nil Nil	56 200 320 520 832	33.6 57.6 91.2 96.0 144.0	440 400 360 300 130	32 26 25 22 18	84.0 68.3 55.6 42.5 17.0	16.19 9.06 6.42 4.45 1.53
······			FFFF			ТАВ	LE—III B							
8.6 6.3 5.6 3.7 2.4	1.65 2.13 2.22 2.76 5.8	200 180 170 153 110	1248 1128 968 820 700	2000 2100 2100 2200 2300	300.0 600.6 610.0 842.0 1120	75 70 65 50 35	Trace Nil Nil Nil Nil	180 50 30 Nil Nil	200 360 460 626 940	57.6 81.6 86.4 100.8 124.8	360 300 280 240 140	30 26 23 18 14	65.6 50.0 43.8 33.9 17.3	8.13 5.22 4.44 3.30 1.61
	2 9.4 7.6 7.0 6.0 2.8 8.6 6.3 5.6	Electrical con- ductivity E.C. Milimohs/cm 2 3 9.4 1.59 7.6 2.13 7.0 2.22 6.0 2.49 2.8 4.20 8.6 1.65 6.3 2.13 5.6 2.22 3.7 2.76	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	HElectrical con- ductivity E.C.Dissolved solidsMilimohs/cmmg/lmg/lmg/l234569.41.5922012321600 \checkmark 7.62.13190107217507.02.22190100018006.02.49180100018002.84.2013057220008.61.65200124820006.32.13180112821005.62.2217096821003.72.761538202200	HElectrical con- ductivity E.C.BOD5CODDissolved solidsChlorides (as Cl)Milimohs/cmmg/lmg/lmg/lmg/l2345679.41.5922012321600 \checkmark 276.97.62.1319010721750454.47.02.2219010001800532.56.02.4918010001800646.12.84.20130572200010658.61.6520012482000300.06.32.1318011282100600.65.62.221709682100610.03.72.761538202200842.0	HElectrical cont ductivity E.C. BOD_5 COD solids(as Cl)(as SO_4)Milimohs/cmmg/lmg/lmg/lmg/lmg/lmg/l23456789.41.5922012321600 \checkmark 276.9657.62.1319010721750454.4757.02.2219010001800532.5606.02.4918010001800646.1502.84.201305722000106535TAB8.61.6520012482000300.0756.32.1318011282100600.6705.62.221709682100610.0653.72.761538202200842.050	HElectrical con- ductivity E.C.BOD5CODDissolved solidsChlorides (as Cl)Sulphates (as SO4)Carbonate (as CaCo3)Milimohs/cmmg/lmg/lmg/lmg/lmg/lmg/lmg/lmg/l234567899.41.5922012321600 \checkmark 276.96557.62.1319010721750454.475Nil7.02.2219010001800532.560Nil6.02.4918010001800646.150Nil2.84.201305722000106535NilTABLE—III B8.61.6520012482000300.075TraceTABLE—III B8.61.6520012482000600.670Nil5.62.221709682100610.065Nil3.72.761538202200842.050Nil	HElectrical con- ductivity E.C.Dissolved SOIDChlorides solidsSulphates (as Cl)Carbonate (as SO_4)Br-Carbonate (as CaCo3)Milimohs/cmmg/lmg/lmg/lmg/lmg/l(as CaCo3)23456789109.41.5922012321600 \checkmark 276.96552507.62.1319010721750454.475Nil1807.02.2219010001800532.560Nil106.02.4918010001800646.150NilNil2.84.201305722000106535NilNilTABLE—III B8.61.6520012482000300.075Trace1806.32.1318011282100600.670Nil505.62.221709682100610.065Nil303.72.761538202200842.050NilNil	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $	$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	$\begin{array}{c c c c c c c c c c c c c c c c c c c $

TABLE—III A. CHARACTERISTICS OF WASTEWATERS AFTER MIXING GRADE III EFFLUENT WITH CHLORINATION & HYPOCHLORITE STAGE BLEACH PLANT EFFLUENT IN DIFFERENT PROPORTIONS

Percent Na = $\frac{Na \times 100}{Na + K + Ca + Mg}$

All cations expressed in terms of meq/1.

= $\overline{/Ca + Mg}$

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S.A.R.

All cations expressed in terms of meq/1.

Grade III Effluent + Chlorination	pН	E.C.	BOD₅	COD	Dissolved solids	Chlorides (as Cl)	Sulphate (as SO ₄)	Carbonate (as CaCO3)	Bicarbonate (as CaCO3)	Calcium (as Ca)	Magnesium (as Mg)	Sodium (as Na)	Potasium (as K)	Na	S.A.
Hypochlorite effluent mg/l milimohs/cm	mg/l mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	mg/l	%			
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	10
$ \begin{array}{c} 1 : 0 \\ 3 : 1 \\ 2 : 1 \\ 1 : 1 \\ 1 : 0 \end{array} $	8.8 7.1 6.95 5.3 2.55	1.56 2.19 2.29 2.64 5.60	250 210 190 180 125	1552 1208 1144 1008 744	1860 1900 1900 1900 2000	269.8 525.4 603.5 781 1313.5	110 70 65 60 45	7.0 Nil Nil Nil Nil	305 200 160 40 Nil	104 320 400 560 1040	14.4 24.0 33.6 48.0 110.4	460 400 350 310 160	28 26 22 20.6 16	83.6 64.2 54.7 45.0 18.4	15. 8. 6. 4.
							TABL	E—III D		<u></u>					
$ \begin{array}{r} 1 : 0 \\ 3 : 1 \\ 2 : 1 \\ 1 : 1 \\ 0 : 1 \end{array} $	9.0 8.2 8.0 7.2 3.5	2.07 2.10 2.25 2.43 3.30	360 310 290 260 130	2163 1450 1348 1167 528	2000 1900 1800 1800 1700	440.2 568.0 582.2 681.6 1022.4	95 85 70 60 38	10 Nil Nil Nil Nil Nil	400 330 290 210 Nil	112 288 360 440 584	38.4 130 153 211 288	530 466 353 300 160	35 30 26 25 20	81.2 60.2 48.7 38.89 20.5	15. 8. 5. 4. 1.