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Silica enters the chemical recovery system in a paper mill through bamboo (1.5-2.5% SiO₂ o.d.) and lime (5-15% SiO₂). Silica will be present as sodium silicate in black liqour and green liquor and as calcium silicate in lime mud. In bamboo based paper mills the general range of silica concentration will be 0.5-2.5, 0.5-12.0, 7.0-43.0 and 9.0-14.0 $g SiO_{9}/l$ for white liquor, weak black liquor, concentrated black liquor and green liquor respectively; concentration of silica in lime sludge will vary over the range 5-15%. The presence of high concentration of silica in lime sludge poses problems during reburning in a kiln for lime recovery. Silicate would melt at the usual calcination temperature levels in the kiln (1100-1300°C) and tend to seal the pores of the lime pellet by forming a glasslike coating enclosing a pellet with an overburned outer layer and an inner core of incompletely calcined lime. Partially calcined lime pellets from silica rich lime

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Conservation of Lime by Desilication of Kraft Green Liquor

Carbonation of green liquor at $60-80^{\circ}$ C with flue gas to pH=9.5 will remove 80-90% of silica in the liquor. Causticization of desilicated green liquor will give a lime sludge $(1.5-2.5\% SiO_3)$ suitable for reburning in a kiln. Reuse of reburned lime will reduce fresh lime requirement by 90-95% compared to present practice of once through use of lime. Preliminary design and cost estimates for a proposed green liquor desilication process favour further work on a pilot plant scale in a paper mill.

sludge will have less available calcium oxide and also give slow and incomplete slaking. Paper mills based on bamboo do not have lime reburning facilities because of the above problems. Instead lime is used in oncethrough operation. Fresh lime stone is calcined in a country kiln or rotary kiln and used for causticization. The resulting lime sludge rich in silica (5-15% SiO₂) is disposed off Disposal of lime sludge (0.3–0.7 ton/ton of pulp) is a major and costly solid waste disposal problem at many plant sites.

Silicate in black liquor also deposits as a hard scale in digester walls and evaporator tubes and leads to beehive fomation in recovery boiler. Sodium silicate in green liquor decreases causticizing efficiency. Thus desilication of black liquor or green liquor would reduce the silica problem and also permit satisfactory calcination of lime and (1.5 - 2.5%)SiO₂) for reuse. Reburning of lime would reduce fresh lime requirement by 90-95%. Sodium silicate in alkaline process liquors can be precipitated by acidification as free silica and as silicate magnesium by calcium oxide, sulfate and aluminum sulfate. This investigation deals with desilication of green liquor by carbonation for tackling the silica problem at the green liguor stage.

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Proposed Green Liquor Desilication Process:

Green liquor contains sodium carbonate, sodium sulfide, sodium hydroxide, sodium sulfate and Silica sodium silicate. can be precipitated from green liquor by carbonation with flue gases containing carbon dioxide from either the recovery boiler or lime kiln. During carbonation, the following principal reactions take place, equation (1,2,3).

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$$CO_{s}+OH^{-} \Leftrightarrow HC_{s}^{-}$$

$$CO_{s} + CO_{s}^{-} + H_{s}O \Leftrightarrow 2HCO_{s}^{-}$$

$$SiO_{s}^{-} + 2HCO_{s}^{-} \xleftarrow{} 2CO_{s}^{-} + Sio_{s} + H_{s}O$$

$$HS^{-} + HCO_{s}^{-} \xleftarrow{} H_{s}S + CO_{s}^{-}$$

Absorption of carbon dioxide by green liquor followed by reactions (1) and (2) decrease the pH of the liquor. Precipitation of silica takes place according to reaction (3). Carbon dioxide is released from carbonated green liquor by raising the liquor temperature to favour the reversal of reaction (2). Experimental studies using commercial and laboratory green liquor samples have shown that 80-90% of silica in green liquor can be removed by 60-86°C carbonation to at pH=9.5 with subsequent settling and filtration. The pH of carbonated green liquor should not dccrease below 9.5; since reaction (4) becomes predominant below pH=9.5 and leads to sulfur loss from system as hydrogen sulfide. Temperature determines the nature of the silica precipitate which governs settling and filtration rates; silica separates as a gel below 44-46°C and as easily filtrable granules above 50°C during carbonation.

The results of laboratory experiments dealing with desilication efficiency and settling and filtration characteristics of the silica precipitate are incorporated in a proposed process¹. Figure 1 gives a plausible flow diagram of

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(1) with the balance for the proposed desilication process.

(2)

(3)

(4)

the proposed process. The follo-

wing additional units are required

for the desilication of green

liquor with flue gas carbonation.

1. Turbulent contact absorber

for carbonation of green

liquor by flue gas (absorber

consists of hollow poly-propy-

lene spheres maintained in a

fluidized state between retai-

ning grids for handling slur-

slurry concentration in carbo-

2. Thickner for increasing the

3. Filter for the separation of

4. Steam stripping column for

5. Kiln for reburning of lime

Special Features of Proposed

A summary of flow rates and

compositions of themajor streams

in the proposed desilication pro-

cess is given in Table 1. Material

balance for silica in various pulp

mill streams is given in Table 2

for a hypothetical 250 ton per-

day bamboo kraft pulp mill along

decomposing bicarbonate in

carbonated green liquor for

causticization,

nated green liquor.

silica precipitate.

satisfactory

and

Process :

mud.

ries).

The concentration of silica in green liquor decreases from 7.25 to 0.82 g SiO₂/1 with the proposed desilication process which results in a lime sludge containning 1.2% SiO₂ compared to 5.7% SiO₂ in the conventional process. The concentration of liquor also silica in white decreases from 2.9 g SiO₂/1 to 0.15 g SiO₂/l and this in turn gives a marginal decrease in the silica concentration of black liquor and green liquor streams over the values in conventional recovery plant.

A summary of design specifications and preliminary cost estimates for the units in the proposed desilication process is given in Table 3. The total fixed capital investment for the process will be Rs. 45.0 lakhs. Operating costs and saving from the modified process are given below :

Operating cost (daily basis) : Utility and Power Fuel oil 22 t @ Rs. 22,000 Rs. 1000/t Makeup lime 6 t @ Rs. 1,800 **Rs. 300/t** 600 Steam 10 t @ Rs. 60/t Rs. Power, 3000 kWh @ 600 Rs. Rs. 0.2/kWh Labour, supervision 650 Rs. and maintenance Depreciation, taxes **Rs.** 1,380 and insurance, etc. Total operating cost Rs. 27,030 Savings with desilication =(Fresh lime consumption

=(Fresh lime consumption with no desilication+

Table 2—Material Balance for Silica In A 250 Ton Per Day Bamboo Kraft Pulp Mill*

Iaput	54		Output			
	Recovery	Process**		Recovery process		
Source	Conven- tional	Madified	Source	Conven-	Modified	
Bamboo chips (2.1%	11350	113 0	Scales:			
White liquor	3920	172	Digester	168	168	
		a Artas	Blow tank	and		
makeup sait cake	160	160	Storage ta	nks 504	410	
Ench lime (C (0/)			Evaporato	r 2500	2000	
$\mathbf{Fresh} \text{lime} (6, 5_{6})$	7330	450		450	350	
Dohumod Hafe (1 00	No.	1. A.	Lime kiln		202	
Reburned nme (1.87	() —	2040	Pulp	1590	1290	
	a sa angan		Black lique	or losses	300	
4.	-		Slaker grit	s 3400	1000	
	B. C. Start		White liqu	or 3920	172	
A CARLER AND			Lime mud	10228	<u>`</u>	
and the second	la de la composición de la composición Este de la composición	en de la	Reburned	lime —	2040	
	and		Silica bypr	oduct	6240	
Total	22760	14172		22760	14172	

*kg SiO₂/day

** Conventional recovery process disposes sludge Modified recovery process removes 85% of silica in green liquor by carbonation and lime sludge is calcined and reused.

Table 3

Design Specifications And Cost of Units In Proposed Desilication Process

TT-	in One of Contraction of Contractio	·	
Un	it Specifications.		Cost. Rs. (Lakhs)
1.	Turbulent contact absorber		(<u>Lukils</u>)
	Diameter-09 m	1. th	4. V
	Number of stages-4		
	Total height-6.4 m		
	Packing 35 mm hollow	n alaa	en di Agenti di
	spheree	polypropylene	
2	Thickener*		and the second sec
	Diamatan 11.5		n a statu 👘 🛶
	Diameter - 11.5 m	14 N 1	
<u>``</u>	Height = 3.0 m	and the second	
3.	Steam Stripper	de la companya de la	5 ()
•	Diameter-1.0 m		· · · · · · · · · · · · · · · · · · ·
	Height —15.0 m		
	Packing-50 mm polypro	nvlene nall ringe	1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 - 1995 -
4.	Filter	bliene ban 1mgs	
	Diameter 1 8 m	190	2.0
~	Length -1.2 m		Constant Constant of Constant
	Filter area 7.0		40.
5	Rotary kiln	· · · · · · · · · · · · · · · · · · ·	
5.	Diamatan 2.77	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	16.0
c ³⁴	Auriliana 6 allia		
o.	Auxiliary facilities, pumps,	· · · · · · · · · · · · · · · · · · ·	18.0
	pipelines, insulation, inst	rumentation	
¥.,	and construction cost		
t des	Total fixed capital investi	ment	180
			‴ 4 ⊅.⊍ ,

*Green liquor clarifier can be used as a thickner and green liquor sent from smelt dissolver to carbonator. silica byproduct+lime mud disposal cost-operating cost for desilication) = $(113 \times 300 + 6.2 \times 200 + 290 \times 25 - 27030)$ =Rs. 15360

(Fresh lime=300/ton, silica byproduct=200/ton, Cost of lime mud disposal=Rs. 25/ton)

Saving per ton of pulp=Rs. 60

Desilication of green liquor by carbonation at 80°C to pH=9.5 will remove 80-85% of silica in liquor and improve causticization efficiency and white liquor clarity. The concentration of silica in lime mud will decrease to 1-2% SiO₂ and will allow satisfactory calcination in a rotary kiln. Reuse of reburned lime will reduce fresh lime requirement by 90-95% compared to present practice. Preliminary design and cost analyses favour further development work on pilot plant scale.

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Fig. 1 Proposed process for silica removal from green liquor,

<u></u>	Summary of	Liquor Compositions And Flow Rates In The Modified Chemical Recovery Process Flow rate h-1 Composition, g/1											
51. No.	Description	Volume m ³	Solids kg	Water kg.	Total kg.	NaOH	NaHS	Na ₂ S	Na ₂ Co ₃	NaHCO ₃	Na ₂ SO ₄	Na ₂ SiO ₃	SiO ₂
1.	Green liquor	58.15	8445	55450	63895	13.6		19.4	99.2		1.80	11.40	5.60
2.	Carbonated green liquor	59.6	10480	55000	65480		12.63	1.9	94.5	64.4	1.80	1.67	0.78
3.	Silica	<u> </u>	260	166	. 426		—	· <u> </u>		—			
4.	Carbonated green liquor to stripper	60.7	10480	56150	0 6663	0 -	12.4	3 1.8	92.8	63.3	1.75	1.64	0.81
5.	Stripped green liquor	59.7	9045	56500	65545	; —	12 62	1.9	134.8	_	1.75	1.67	0.82
6.	White liquor	47.7	5870	45600	51470	80 .	4 -	19.6	21.2		1.75	0.51	0.15

TABLE 1

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